

Original Research Article

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## A Comparative Study between Three Aquatic Plants to Phytoremediation of Lead from Wastewater

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### ABSTRACT

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This study was conducted in vitro to identify the three plants the ability of a *Ceratophyllum demersum* and *Hydrilla verticellata* and *Echhornia crassipes* to remove the lead element of the sewage using different dilutions (25,50, 100)% of wastewater for a period of thirty days each plant separately and compare the results showed that the three plants have the ability to remove lead from sewage and accumulation in their tissues was *C.demersum* removes bullets in the leg above the securities while *H.verticellata* plant accumulates lead in the root <paper <leg while leaf *E.crassipes* was has the ability to accumulate lead in the root <paper, as compared to the three plants in its ability to remove lead in the *E.crassipes* interweave appeared leaf ultra removal of lead capacity. The concentration of elements in the tissues in large quantities. And it found significant differences between the parts of the plant at the level of the probability  $P \leq 0.05$  in most of concentrations studied during the duration of the experiment, recorded significant differences among the three plants.

### Introduction

Multiple wastewater Sources, there is domestic sewage, industrial waste and divert river water and as a result of increased industrial and agricultural development and increasing human activities led to an increase wastewater which requires wastewater treatment to achieve environmental sustainability.

The wastewater, one of the sources of environmental pollution in general and contamination of water resources is particularly so when put into rivers, seas and

lakes without treatment and consequences of this it causes to all the health, agricultural and tourist areas which requires wastewater treatment before putting numerous methods of wastewater treatment but most are expensive and far for optimal performance to them. But the use of phytoremediation is the best method because it is inexpensive and environmentally friendly (Memon *et al.*, 2001) and the means to use some of the plants that have the ability to reduce the level of contamination by certain mechanisms carried out by the plant lead to

the removal or analysis of different contaminants (Sasaki *et al.*, 2003). And that the processing plant is one of the vital technologies for green plants that are used to treat contaminants in the environment expel extracellular toxicity or reduce movement in the soil or water and at increasing the level of heavy metals within the plant tissue, the plant but that is compiled in special locations in the root or leg or converts to other forms non-toxic as possible be distributed and used again in the metabolic processes (Maine, *et al.*, 1999).

Many plants can accumulate contaminants in their tissues and rid the environment of different minerals, some of which are essential for plant growth as key elements in the growth such as iron, manganese, zinc, copper, magnesium, nickel and others accumulates in the plant body, but an element is essential such as cadmium, lead, mercury and carbon source (Qian *et al.*, 1999).

The accumulation of mechanical elements within the plant body is the fact that these toxic elements linked to the walls of the cells in roots or leaves which prevents they move through the wringer plant. or expel by mechanical especially to the non-sensitive sites in the cell as it is stored in the gaps (Hettiarachchi *et al.*, 2012).

### **Phytoremediation**

The term of phytoremediation comes from a Greek prefix “Phyto” and attached to latin word “remediation” to correct. Simply, it refers to utilize green plants to remove contaminants from soil and water (Mitra, 2012). It is cost effective, as ethically accepted, and it is less disturbed to the environment. The plants used in this ecofriendly technology must have a considerable capacity of metal uptake. There

are many ways by which plants remediate contaminated sites.

### **Phytoextraction**

Phytoextraction is the uptake of contaminants by roots and trans-located into aerial parts, which enrich with metals. To reach the acceptable level of metals, plants should use continuously. However, phytoextraction is among the best approach to remove contaminants from soil, sludge and sediment, it takes a long time (Malik and Biswas, 2012).

### **Rhizofiltration**

Rhizofiltration is the uptake and concentrate (absorb and adsorb) contaminants by roots from pollutedarouses sources. The tolerance and translocation of the metals to aerial parts are irrelevant (Jadia and Fulekar, 2009).

### **Phytovolatilization**

In phytovolatilization, plants use to uptake contaminants from, soil, sediments and sledges transforming them into volatilized compounds after that, transpiring them into the atmosphere (Prasad, 1998).

### **Phytostabilization**

It is defined as the use of transcription and root growth to reduce the mobility bioavailability of pollutants in the environment through creating an aerobic condition in the root zone and could be achieved via the addition of organic matter and soil (Hussner and Losch, 2005).

Heavy metals removed by the plants due to storage in the roots, stems, or leaves. The metals convert into less harmful substances within the plant or gaseous form and are released into the air through transpiration activities (Alluri *et al.*, 2007).

## Materials and Methods

### Collection of plants

Three types of aquatic plants collected represented by (*C. demersum*, *H. verticellata*, and *E. crassipes*) and samples of water from the Tigris River in the Jadiriya district of Baghdad province (Figure1).

**Figure.1** Show precipitation basins Elementary in Rustumiya station in Baghdad



After three plants collected in plastic bags with a little water from the same site and transported to the laboratory as it was washed with fresh tap water and a small brush to remove sediment, aquatic invertebrates while preserving the root system.

Almost similar-sized plants planted in the age and stage in bathtubs glass of Education with the dimensions of 20 × 30 × 40 cm, taking into account the volume of water found in ponds and then compensate the lost water, plants and distributed on four basins in order to avoid crowding and competition for the place.

And it has acclimatization of plants for the temperature of the laboratory 24°C was placed 10 liters of free chlorine water (up to 24 hours with exposure to the sun) in ponds

until the test after it measured the proportion of lead element in the roots and leaves of plants and a sample of water of the river also will be remembered later.

### Collection of Wastewater

Wastewater samples collected in the health of the liquidation Rustumiya station's picture (1) located in the Rustumiya region in the province of Baghdad, was the sampling of primary sedimentation tanks periodically quarterly and then added to chloroform to prevent any biological activity then transferred the samples by plastic containers clean to the lab where it was placed in the refrigerator degree 16:00 until a required analysis by the policies included in the standards 1995, Methods).

### Preparation Wastewater Dilutions

Proper dilutions needed for wastewater with three replicates for each dilution, which was (25% -50% -100%) of sewage water have been added to the tubs with dimensions of 20 \* 20 \* 30 cm experience which nine 4-liter, and put the plants in the basin after the measure concentrations of heavy metals in the leaves and roots and then make the necessary measurement determined by atomic spectrometer Flame steps Flame atomic Absorption Spectrophotometer type (VGP 210 Buck, England) depending on (Senila *et al.*, 2011)

### Heavy Metals Measurement

After making dilutions necessary element lead with three replicates for each concentration were added to the tubs experience with the dimensions of 20 \* 20 \* 30 cm containing 4 liters of distilled water, taking into account the development of needed nutrients, as it put the plants in the basin after the measurement of the concentration of heavy elements in the

leaves and roots and then make the necessary steps to measure determined by atomic spectrometer Flame Atomic Absorption Spectrophotometer type (VGP 210 Buck, England) depending on (Senila et al., 2011), as follows: Measurement of heavy elements in the leaves, stems and roots.

Collect 10 grams of leaves, stems and roots placed in a bottle an hour in an electric oven 105 ° degree m until the drought separately.

Weigh 1 g of dry specimen of both the leaves and roots in flasks glass Beakers and added a 12 ml of nitric acid Center HNO<sub>3</sub> (64%) to conduct the process of digestion (Digestion) and then add 4 ml of hydrogen peroxide H<sub>2</sub>O<sub>2</sub> (30%). Preheat the specimen in the oven Oven MEMMERT ISO 9001, GERMANY)) at a temperature of 120 ° C for a period of 2 hours.

Lift the specimen from the oven and let cool for 24 hours and added a 10 ml of distilled water. Nominated mediated by the nomination of the type Millipore paper (Whatman 0.45µm) and the resulting solution placed in a glass cylinder (Salender) 50 ml size and he finished with distilled water to the mark. Reference Blank attended the solutions in the same way without the use of a specimen.

It been estimated concentrations of elements in the models mediated by a flame atomic absorption spectrum Flame Atomic Absorption Spectrometry Model (VGP 210 Buck, England) English-made.

### **Measurement of Heavy Metals in the Water**

Measure the amount of 150 ml of water in ponds and well posts On was nominated in

the volumetric flask 500 mL. Added per 100 ml of the specimen 2 ml of nitric acid HNO<sub>3</sub>. Attended lotions reference Blanks same way.

The estimation elements determined by atomic spectrometer Flame mentioned above (Dumitrescu & Arpad, 2013).

### **Results and Discussion**

Comparison of the three plants growing in the basins of wastewater in their ability to accumulation of lead Shapes (2, 3,4,5 and 6) can be observed comparing the results between the three aquatic plants (*E.crassipes*, *H.verticellata*, *C.demersum*) developing in the basins of wastewater in their ability to accumulation of lead in tissues showed plant *E.crassipes* has the ability to accumulation of lead to interweave more than vegetarian *H.verticellata*, *C.demersum* as accumulated leaf water hycianth (0.453, 0.749, 0.885) ppm concentrations (100,50,25) % of the wastewater in the age of thirty-day experiment as shown in the table (18).

While for the plant *H.verticellata* wilted and died at the concentration of 50% of the wastewater in the fifth day of life experience, and it was the highest accumulation of his first day at the focus 100% of the wastewater 0.448ppm, either for the plant *C.demersum* has died at a concentration of 100% the tenth day of life experience and reached the highest accumulation of him when the concentration of 50% in the tenth day of life experience. And reach through these results to be the order of the plants in the ability to remove lead from wastewater is *E.crassipes* <*C.demersum* <*H.verticellata*.

**Table.1** Compared to the concentrations of lead in plants *E.crassipes*, *H.verticellata*, *C.demersum* after different treatment concentrations Of wastewater is expressed in units of ppm by different times

LSD	Concentrations - ppm			Plants	Time of Experimental
	100%	50%	25%		
0.145 *	0.362	0.177	0.0651	<i>Ceratophyllum demersum</i>	First day
0.209 *	0.448	0.196	0.088	<i>Hydrilla verticellata</i>	
0.182 *	0.498	0.244	0.097	<i>Echhornia crassipes</i>	
-	0.206 NS	0.167 NS	0.044 NS	LSD	
0.183 NS	-	0.339	0.165	<i>Ceratophyllum demersum</i>	Fifth day
0.264 *	-	0.745	0.204	<i>Hydrilla verticellata</i>	
0.372 *	0.832	0.445	0.201	<i>Echhornia crassipes</i>	
-	-	0.215 *	0.133 NS	LSD	
0.204 *	-	0.544	0.239	<i>Ceratophyllum demersum</i>	Tenth day
-	-	-	0.343	<i>Hydrilla verticellata</i>	
0.467 *	1.122	0.605	0.317	<i>Echhornia crassipes</i>	
-	-	0.228 NS	0.172 NS	LSD	
-	-	-	-	<i>Ceratophyllum demersum</i>	Fifteenth day
-	-	-	-	<i>Hydrilla verticellata</i>	
0.351 *	0.367	0.802	0.397	<i>Echhornia crassipes</i>	
-	-	-	-	LSD	
-	-	-	-	<i>Ceratophyllum demersum</i>	Thirteenth day
-	-	-	-	<i>Hydrilla verticellata</i>	
0.285 *	0.749	0.885	0.453	<i>Echhornia crassipes</i>	
-	-	-	-	LSD	
NS:Not significant			* P<0.05		

Fig.2 Map of Baghdad described the site to collect samples



Fig.2 Compared to the concentrations (ppm) of lead in plants *E.crassipes*, *H.verticellata*, *C.demersum* after One day of experiment period.

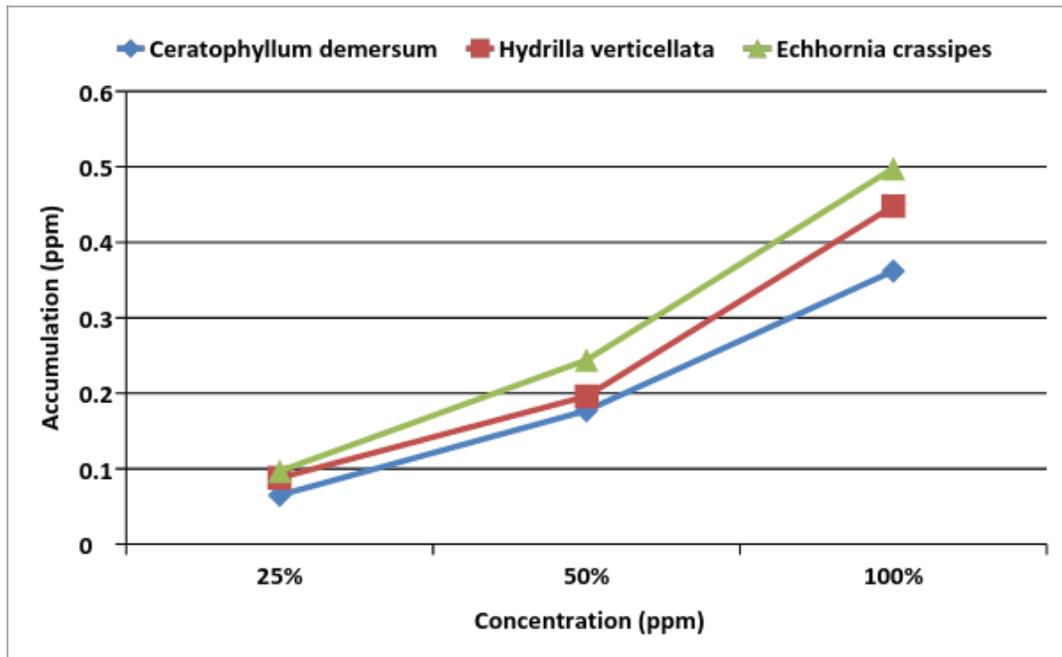


Fig.3 Compared to the concentrations (ppm) of lead in plants *E.crassipes*, *H.verticellata*, *C.demersum* after Five days of experiment period

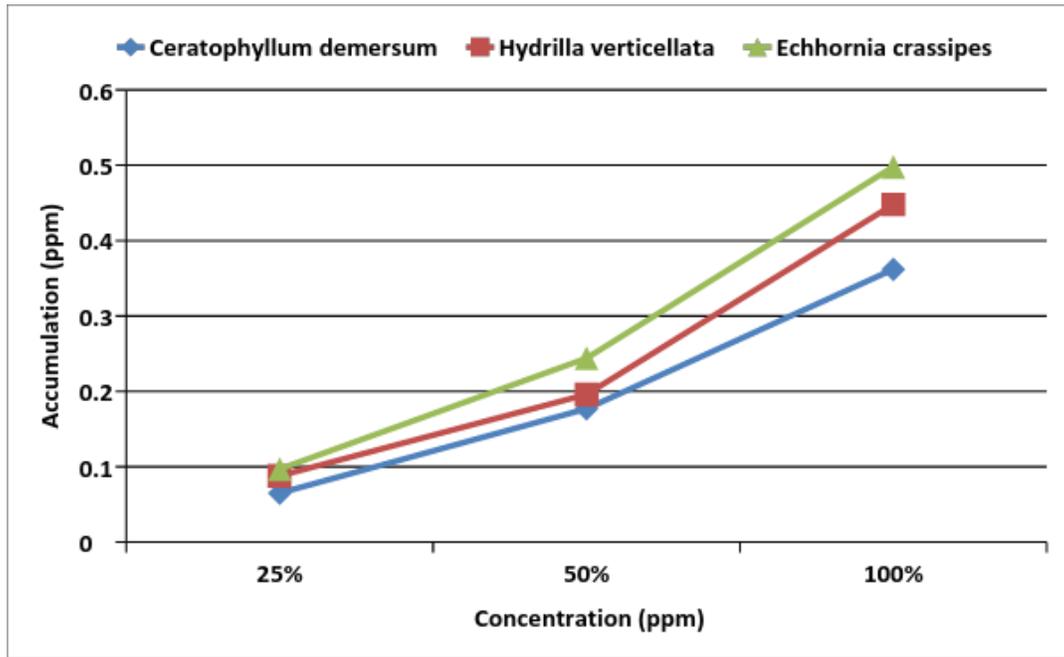


Fig.4 Compared to the concentrations (ppm) of lead in plants *E.crassipes*, *H.verticellata*, *C.demersum* after Ten days of experiment period.

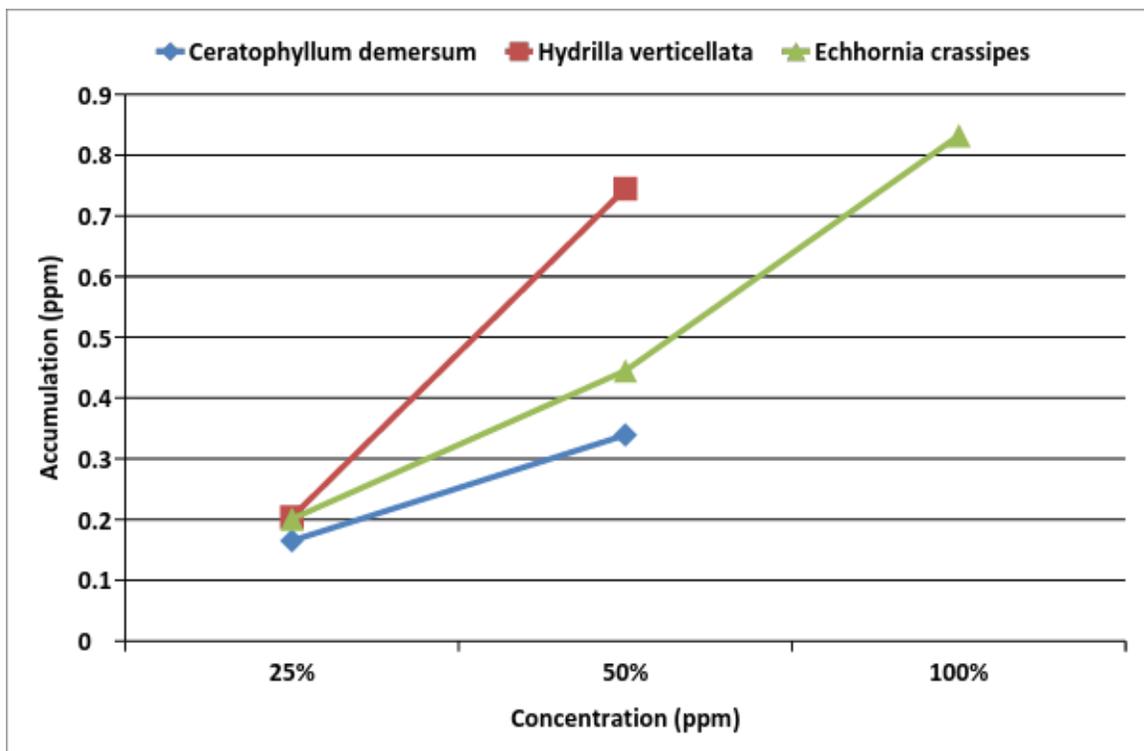


Fig.5 Compared to the concentrations (ppm) of lead in plants *E.crassipes*, *H.verticellata*, *C.demersum* after fifteen days of experiment period

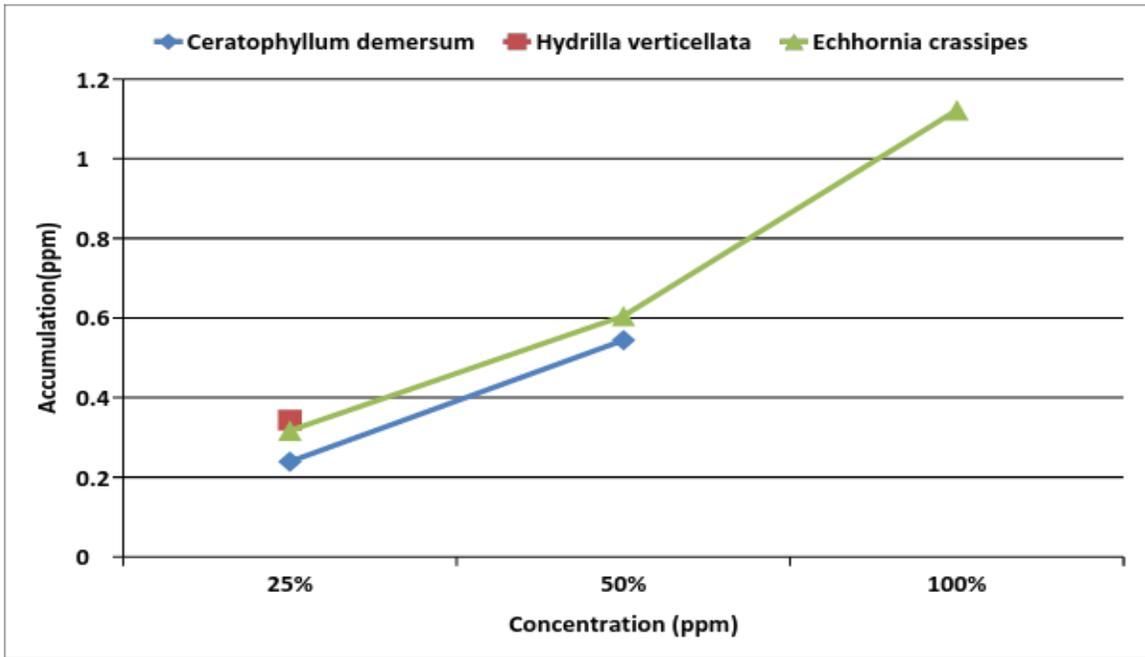
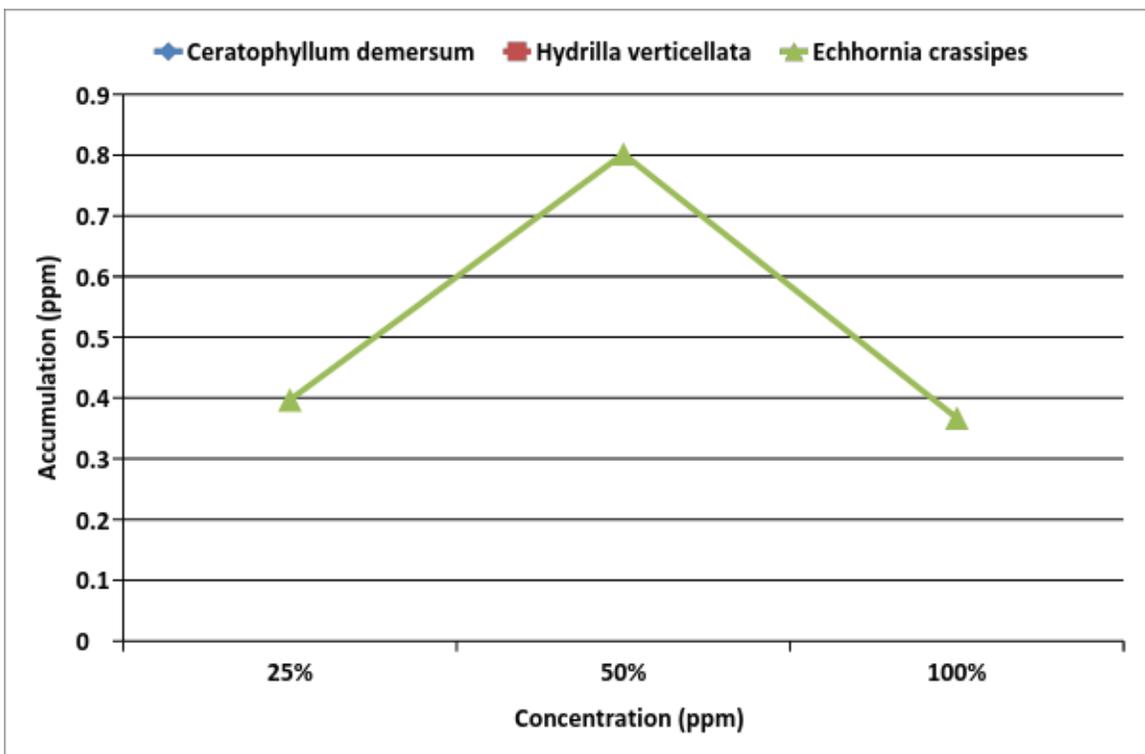


Fig.6 Compared to the concentrations (ppm) of lead in plants *E.crassipes*, *H.verticellata*, *C.demersum* after thirty days of the experiment of experiment period



Results of statistical analysis showed that there were significant differences at concentrations (25.50%) in the fifth and tenth days of life experience and the results showed no significant differences at all three concentrations in the first and fifth of lifetime experience. Consistent results of this study with (Soltan and Rashed, 2003) that the plant *E.crassipes* removes effectively clear amounts of metals Pb and Ni and Mn and Cu and Cr and Co and Cd and Zn of fresh water, especially in the concentrations of low-lying and that the economic success of Operation phytoremediation depends largely on the effectiveness of photosynthesis and the photosynthetic rate of growth rate of the plant.

Due to the fact *E.crassipes* high growth speed and productivity, which made them effective in the treatment of types of contaminated water (Xia and Ma, 2006). Found that *E.crassipes* capable of good growth in the waste water of dairy factories as it reaches the value of the chemical oxygen demand of more than 1160 ppm.

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